Molecules as Automata

Representing Biochemical Systems as Collectives of Interacting Automata

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Macro-Molecules as Interacting Automata

Cells Compute

- No survival without computation!
 - o Finding food
 - Avoiding predators
- *How* do they compute?
 - Unusual computational paradigms.
 - Proteins: do they work like electronic circuits?
 - Genes: what kind of software is that?

• Signaling networks

- Clearly "information processing"
- They are "just chemistry": molecule interactions
- But what are their principles and algorithms?
- Complex, higher-order interactions
 - MAPKKK = MAP Kinase Kinase Kinase: that which operates on that which operates on that which operates on protein.
- General models of biological computation
 - \circ What are the appropriate ones?



<u>cascade</u>, Chi-Ying F. Huang and James E. Ferrell, Jr., 1996, <u>Proc. Natl. Acad. Sci. USA</u>, 93, 10078-10083.

Biological "Algorithms"





Viral Replication



Adapted from: B.Alberts et al. Molecular Biology of the Cell third edition p.279.





H.Lodish et al. Molecular Cell Biology. Cardelli fourth Edition p.730.

Discrete State Transitions



Compositionality (NOT!)



Process Algebra

- Reactive systems (living organisms, computer networks, operating systems, ...)
 - Math is based on *entities that react/interact with their environment* (*"processes"*), not on functions from domains to codomains.
- Concurrent
 - Events (reactions/interactions) happen concurrently and asynchronously, not sequentially like in function composition.
- Stochastic
 - Or probabilistic, or nondeterministic, but is never about deterministic system evolution.
- Stateful
 - Each concurrent activity ("process") maintains its own local state, as opposed to stateless functions from inputs to outputs.
- Discrete
 - Evolution through discrete transitions between discrete states, not incremental changes of continuous quantities.
- Kinetics of interaction
 - An "interaction" is anything that moves a system from one state to another.



Kinetic laws:



Kinetic laws:

Two complementary actions may result in an interaction.



Kinetic laws:

Two complementary actions may result in an interaction. A decay may happen spontaneously.

Interacting Automata Transition Rules



Q: What kind of mass behavior can this produce?

(We need to understand that if want to understand biochemical systems.)



The equivalent process algebra model



Interactions in a Population



Interactions in a Population



Interactions in a Population



Interactions in a Population (2)



Interactions in a Population (2)



CTMC Semantics





CTMC

Stochastic Collectives

• "Collective":

- A large set of interacting finite state automata:
 - Not quite language automata ("large set")
 - Not quite cellular automata ("interacting" but not on a grid)
 - Not quite process algebra ("collective behavior")
 - Cf. multi-agent systems and swarm intelligence

• "Stochastic":

- o Interactions have rates
 - Not quite discrete (hundreds or thousands of components)
 - Not quite continuous (non-trivial stochastic effects)
 - Not quite hybrid (no "switching" between regimes)
- Very much like biochemistry
 - \circ Which is a large set of stochastically interacting molecules/proteins
 - Are proteins finite state and subject to automata-like transitions?
 - Let's say they are, at least because:
 - Much of the knowledge being accumulated in Systems Biology is described as state transition diagrams [Kitano].



Chemistry vs. Automata



Groupies and Celebrities

Groupies and Celebrities



A stochastic collective of celebrities:



Stable because as soon as a A finds itself in the majority, it is more likely to find somebody in the same state, and hence change, so the majority is weakened.





Groupies and Celebrities

?a







A stochastic collective of groupies:



Unstable because within an A majority, an A has difficulty finding a B to emulate, but the few B's have plenty of A's to emulate, so the majority may switch to B. Leads to deadlock when everybody is in the same state and there is nobody different to emulate.

Both Together

A way to break the deadlocks: Groupies with just a few Celebrities



Hysteric Groupies

We can get more regular behavior from groupies if they "need more convincing", or "hysteresis" (history-dependence), to switch states.





Some Devices



Ultrasensitive Switch



Cascade Amplifier





More Devices



Repressilator (1 of 3 similar gates)





Design Exercise: Making Lines

Build me a population like this:



Second-order and Zero-order Regime



$E+S \rightarrow^{r} E+P$ directive sample 1000.0 directive plot S(); P(); E() new a@1.0:chan() let E() = !a; E() and S() = ?a; P() and P() = () run (1 of E() | 1000 of S())

Second-Order Regime d[S]/dt = -r[E][S]



Cascades



Second-Oder Regime cascade: a signal amplifier (MAPK) aHi > 0 ⇒ cHi = max

ective sample 0.03
ective plot la; lb; lc
v a@1.0:chan new b@1.0:chan new c@1.0:chan
Amp_hi(a:chan, b:chan) =
o !b; Amp_hi(a,b) or delay@1.0; Amp_lo(a,b)
d Amp_lo(a:chan, b:chan) =
a; ?a; Amp_hi(a,b)
1 1000 of (Amp_lo(a,b) Amp_lo(b,c))
A() = !a; A()
1 100 of A()





Zero-Oder Regime cascade: a signal *divider!*

aHi = max \Rightarrow cHi = 1/3 max

directive sample 0.03
directive plot !a; !b; !c
new a@1.0:chan new b@1.0:chan new c@1.0:chan
let Amp_hi(a:chan, b:chan) =
do lb; delay@1.0; Amp_hi(a,b) or delay@1.0; Amp_lo(a,b)
and Amp_lo(a:chan, b:chan) =
?a; ?a; Amp_hi(a,b)
run 1000 of (Amp_lo(a,b) Amp_lo(b,c))
let A() = !a; delay@1.0; A()
run 2000 of A()

Ultrasensitivity



Zero-Order Regime
A small E-F inbalance causes
a much larger S-P switch.



Second-Order Regime

Design Exercise: Making Waves

Build me a population like this:



Nonlinear Transition (NLT)



Two NLTs: Bell Shape



NLTs in Series: Soliton Propagation





directive sample 0,1 1000

directive plot A1(); A2(): A3(); A4(); A5(); A6(); A7(); A8(); A9(); A10(); A11(); A12(); A13()

val r=1.0 val s=1.0

- new a2@s:chan new a3@s:chan new a4@s:chan
- new a5@s:chan new a6@s:chan new a7@s:chan
- new a8@s:chan new a9@s:chan new a10@s:chan
- new a11@s:chan new a12@s:chan new a13@s:chan
- let A1() = do delay@r;A2() or ?a2; A2()
- and A2() = do la2;A2() or delay@r;A3() or ?a3; A3()
- and A3() = do !a3;A3() or delay@r;A4() or ?a4; A4()
- and A4() = do !a4;A4() or delay@r;A5() or ?a5; A5()
- and A5() = do !a5;A5() or delay@r;A6() or ?a6; A6()
- and A6() = do !a6;A6() or delay@r;A7() or ?a7; A7()
- and A7() = do !a7;A7() or delay@r;A8() or ?a8; A8()
- and A8() = do !a8;A8() or delay@r;A9() or ?a9; A9()
- and A9() = do !a9;A9() or delay@r;A10() or ?a10; A10()
- and A10() = do la10; A10() or delay@r; A11() or ?a11; A11()
- and A11() = do la11;A11() or delay@r;A12() or ?a12; A12() and A12() = do la12;A12() or delay@r;A13() or ?a13; A13()
- and A13() = !a13;A13()

run 1000 of A1()
NLT in a Cycle: Oscillator (unstable)





directive sample 0.03 1000 directive plot A(); B(); C()

new a@1.0:chan new b@1.0:chan new c@1.0:chan let A() = do !a;A() or ?b; B() and B() = do !b;B() or ?c; C() and C() = do !c;C() or ?a; A()

run (900 of A() | 500 of B() | 100 of C())



Oscillator (stable)



Semantics of Collective Behavior

"Micromodels": Continuous Time Markov Chains

- The underlying semantics of stochastic π -calculus (and stochastic interacting automata). Well established in many ways.
 - Automata with rates on transitions.
- "The" correct semantics for chemistry, executable.
 - Gillespie stochastic simulation algorithm
- Lots of advantages
 - Compositional, compact, mechanistic, etc.
- But do not give a good sense of "collective" properties.
 - $\circ~$ Yes one can do simulation.
 - Yes one can do program analysis.
 - Yes one can perhaps do modelchecking.
 - $\circ~$ But somewhat lacking in "analytical properties" and "predictive power".

"Macromodels": Ordinary Differential Equations

- The classical semantics of collective behavior.
 - E.g. kinetic theory of gasses.
 - They always ask: "How does you automata model relate to the 75 ODE models in the literature?"
- Going from processes/automata to ODEs directly:
 - In principle: just write down the Rate Equation:
 - Let [S] be the "number of processes in state S" as a function of time.
 - Define for each state S:

d[S]/dt = (rate of change of the number of processes in state S)
 Cumulative rate of transitions from any state S' to state S, times [S'],
 minus cumulative rate of transitions from S to any state S", times [S].

- Fairly intuitive (rate = inflow minus outflow)
- Going to ODEs indirectly through chemistry
 - If we first convert processes to chemical reactions, then we can convert to ODEs by standard means!



The Two Semantic Sides of Chemistry



These diagrams commute via appropriate maps.

L. Cardelli: "On Process Rate Semantics" (TCS) L. Cardelli: "A Process Algebra Master Equation" (QEST'07)

Quantitative Process Semantics



Stochastic Processes & Discrete Chemistry



Chemical Reactions (FSRN)

Α	\rightarrow^{r}	B ₁ ++ B _n (n≥0) Unary Reaction	d[A]/dt = -r[A]	Exponential Decay
$A_1 + A_2$	\rightarrow^{r}	B ₁ ++ B _n (n≥0	Hetero Reaction	$d[A_i]/dt = -r[A_1][A_2]$	Mass Action Law
A + A	\rightarrow^{r}	$B_1 + + B_n$ (n ≥ 0) Homeo Reaction	$d[A]/dt = -2r[A]^2$	Mass Action Law
				(assuming A≠B _i ≠A _i f	or all i,j)

No other reactions!

JOURNAL OF CHEMICAL PHYSICS VOLUME 113, NUMBER 1 The chemical Langevin equation Daniel T. Gillespie ^{a)} Research Department, Code 4T4100D, Naval Air Warfare Center, China Lake, California 93555 Genuinely trimolecular reactions do not physically occur in dilute fluids with any appreciable frequency. Apparently trimolecular reactions in a fluid are usually the combined result of two bimolecular reactions and one monomolecular reaction, and involve an additional short-lived species.		Chapter IV: Chemical Kinetics [David A. Reckhow , CEE 572 Course] reactions may be either elementary or non- elementary. <u>Elementary reactions</u> are those reactions that occur exactly as they are written, without any intermediate steps. These reactions almost always involve just one or two reactants <u>Non-elementary reactions</u> involve a series of two or more elementary reactions. Many complex environmental reactions are non- elementary. In general, reactions with an overall reaction order greater than two, or reactions with some non-integer reaction order are non-elementary.		THE COLLISION THEORY OF R RATES www.chemguide.co.uk The chances of all this happer your reaction needed a collision involving more than 2 particle remote. All three (or more) pa would have to arrive at exactle same point in space at the sar with everything lined up exact and having enough energy to r That's not likely to happen very	EACTION ning if on s are articles y the ne time, tly right, react. ry often!
Trimolecular reactions: $A + B + C \rightarrow^{r} D$ the measured "r" is an (imperfect) aggregate of e.g.: $A + B \leftrightarrow AB$ $AB + C \rightarrow D$			Enzymatic reactions: $S \xrightarrow{E}^{r} P$ the "r" is given by Mid (approximated steady $E + S \leftrightarrow ES$ $ES \rightarrow P + E$	chaelis-Menten v-state) laws: Luca Cardelli 2009:03-12	45

Chemical Ground Form (CGF)

E ::= 0 : X=M, E M ::= 0 : π ;P \oplus M P ::= 0 : X P π ::= $\tau_{(r)}$: $?a_{(r)}$: $!a_{(r)}$ CGF ::= E,P	Reagents Molecules Solutions Actions (delay, input, output) Reagents plus Initial Conditions		A stochastic subset of CCS (no values, no restriction)
(To translate chemistry to processes we need a bit more than interacting automata: we may have "+" on the right of \rightarrow , that is we may need " " after π .)		 ⊕ is stochastic choice (vs. + for chemical reactions) 0 is the null solution (P 0 = 0 P = P) and null molecule (M⊕0 = 0⊕M = M) Each X in E is a distinct <i>species</i> Each name a is assigned a fixed rate r: a_(r) 	



From CGF to Chemistry





A = $!a;A \oplus ?a;B$ B = $?a;A \oplus \tau_{(s)};A$



 $A = !a;A \oplus ?a;B$ $B = ?a;A \oplus \tau_{(s)};A$



Interacting Automata	Discrete Chemistry
initial states A A A	initial quantities #A ₀
A @r	A → ^r A'
A ?a A' B !a B'	A+B →r A'+B'
?a A !a A' @r A"	A+A → ^{2r} A'+A"



From CGF to Chemistry: Ch(E)

E ::= 0 : X=M, E	Reagents	ΕΥį
M ::= 0 ∶ π;P ⊕ M	Molecules	Å-su
P ::= 0 : X P	Solutions	asso
$\pi ::= \tau_{(r)} : ?a_{(r)} : !a_{(r)}$	Interactions (delay, input, output)	X rea
CGF ::= E,P	Reagents plus Initial Conditions	

E.X.i ≝ the i-th Å-summand of the molecule M associated with the X reagent of E

Chemical reactions for E,P:

(N.B.: <...> are reaction tags to obtain multiplicity of reactions, and P is P with all the | changed to +)

Ch(E) := {(<X.i>: X →^r P) s.t. E.X.i = $\tau_{(r)}$; P} ∪

 $\{(<X.i,Y.j>: X + Y \rightarrow^{r} P + Q) \ s.t. \ X \neq Y, \ E.X.i = ?a_{(r)}; P, \ E.Y.j = !a_{(r)}; Q\} \cup \{(<X.i,X.j>: X + X \rightarrow^{2r} P + Q) \ s.t. \ E.X.i = ?a_{(r)}; P, \ E.X.j = !a_{(r)}; Q\}$

Initial conditions for P:

Ch(P) := P

Entangled vs Detangled





Entangled: Two reactions on one channel Detangled: Two reactions on two separate channels

We need a semantics of automata that identifies automata that have the "same chemistry".

No traditional process algebra equivalence is like this!

Entangled automata lead to more compact models than in chemistry.

Detangled automata are in simple correspondence with chemistry.

Some Syntactic Properties

- C and Ch(Pi(C)) have the same reactions
 - (and their reaction labels are in bijection)
- Def: E is detangled if each channel appears once as ?a and once as !a.
- If C is a system of chemical reactions then Pi(C) is detangled.
 - (hence chemical reactions embed into a subclass of CGFs)
- Hence for any E, we have that Pi(Ch(E)) is detangled.
 - (E and Pi(Ch(E)) are "equivalent" CGFs, but that has to be shown later)
- Def: E,P is automata form if "|" occurs only (other than "|0") in P.
- Def: Detangle(E) is defined from Pi(Ch(E)) by replacing any occurrence pairs ?a_(r); (X|Y|0) and !a_(r); 0 with ?a_(r); (X|0) and !a_(r); (Y|0).
- If E is in automata form then Detangle(E) is (detangled and) in automata form
 o (but Pi(Ch(E)) may not be)

Entangled vs detangled



(closely related to Pi(Ch(E₃)))

Chemical Parametric Form (CPF)

E ::= 0 : X(p)=M, E
M ::= 0 ∶ π;P ⊕ M
P::=0 : X(p) P
$\pi ::= \tau_{(r)} : ?a_{(r)}(\mathbf{p}) : !a_{(r)}(\mathbf{p})$
CPF::= E,P

Not bounded-state systems. Not finite-control systems. But still finite-species systems.

A translation from CPF to CGF exists (expanding all possible instantiation of parameters from the initial conditions)

An incremental translation algorithm exists (expanding on demand from initial conditions)

Reagents

Molecules

Solutions

Actions

with initial conditions

⊕ is stochastic choice (vs. + for chemical reactions)
0 is the null solution (P|0 = 0|P = P) and null molecule (M⊕0 = 0⊕M = M)
Each X in E is a distinct *species*p are vectors of names
p are vectors of distinct names when in binding position
Each free name a in E is assigned a fixed rate r: a_(r)

Example: Neg(a,b) = ?a; Inh(a,b) $\oplus \tau_e$; (Tr(b) | Neg(a,b)) Inh(a,b) = τ_h ; Neg(a,b) Tr(b) = !b; Tr(b) $\oplus \tau_d$; 0 Neg(x,x)

CPF to CGF: Handling Parameters

Consider first the CPF subset with no communication (pure ?a, !a).

Grounding (replace parameters with constants) where X/p is a name in bijection with $\langle X, p \rangle$ (each X/p is seen as a separate *species*) $/(\pi_1; P_1 \oplus ... \oplus \pi_n; P_n) =_{def} \pi_1; /(P_1) \oplus ... \oplus \pi_n; /(P_n)$ $/(X_1(p_1) | ... | X_n(p_n)) =_{def} X_1/p_1 | ... | X_n/p_n$

 $E ::= X_{1}(\mathbf{p}_{1}) = M_{1}, ..., X_{n}(\mathbf{p}_{n}) = M_{n}$ $M ::= \pi_{1}; P_{1} \oplus ... \oplus \pi_{n}; P_{n}$ $P ::= X_{1}(\mathbf{p}_{1}) | ... | X_{n}(\mathbf{p}_{n})$ $\pi ::= \tau_{r} ?a !a$

Let N be the set of free names occurring in E.

E_G is the Parametric Explosion of E (still a finite species system) computed by replacing parameters with all combinations of free names in E

$$\begin{split} & E_G := \{ (X/q = /(M\{p \leftarrow q\})) \text{ s.t. } (X(p) = M) \in E \text{ and } q \in N^{\#p} \} \\ & P_G := /P \qquad \qquad (\text{simply ground the given initial conditions once}) \end{split}$$

 E_G is a CGF! To obtain the chemical reactions $Ch_P(E)$, just compute $Ch_G(E_G)$

 $Ch_P(E) = Ch_G(E_G)$

CPF to CGF: Handling Communication

Grounding (replace parameters with constants)

just one main change: now also convert each input parameter into a ground choice of all possible inputs

N is the set of free names in E,P

#p is the length of **p**

n/p is a name in bijection with <n,p>

X/p is a name in bijection with <X,p>

(each X/p is seen as a separate species)

 E_G is again the **Parametric Explosion** of E

$$\begin{split} & E_G := \{ (X/q = /_N(M\{p \leftarrow q\})) \text{ s.t. } (X(p) = M) \in E \text{ and } q \in N^{\#p} \} \\ & P_G := /_N(P) \qquad \qquad (\text{simply ground the given initial conditions once}) \end{split}$$

 $Ch(E) = Ch_G(E_G)$ E_G is a again a CGF!

 $E ::= X_{1}(\mathbf{p}_{1}) = M_{1}, ..., X_{n}(\mathbf{p}_{n}) = M_{n}$ $M ::= \pi_{1}; \mathbf{P}_{1} \oplus ... \oplus \pi_{n}; \mathbf{P}_{n}$ $P ::= X_{1}(\mathbf{p}_{1}) | ... | X_{n}(\mathbf{p}_{n})$ $\pi ::= \tau_{r} ?a(\mathbf{p}) !a(\mathbf{p})$

CPF to CGF Translation. Ex: Neg(x,x)

E =

Neg(a,b) = ?a; Inh(a,b) $\oplus \tau_e$; (Tr(b) | Neg(a,b)) Inh(a,b) = τ_h ; Neg(a,b) Tr(b) = !b; Tr(b) $\oplus \tau_d$; 0 Neg(x,x)

----- initialization -----

$$E_c := \{ Neg/x, x = ?x; Inh/x, x \oplus \tau_e; (Tr/x | Neg/x, x) \}$$

----- iteration 1 -----

$$C := \{ Neg/x, x \rightarrow^{e} Tr/x + Neg/x, x \}$$

$$E_{c} := \{ Neg/x, x = ?x; Inh/x, x \oplus \tau_{e}; (Tr/x | Neg/x, x) \\ Tr/x = !x; Tr/x \oplus \tau_{d}; 0 \}$$

----- iteration 2 -----

$$C := \{ Neg/x, x \rightarrow^{e} Tr/x + Neg/x, x \\ Tr/x \rightarrow^{d} 0 \\ Tr/x \rightarrow^{d} 0 \}$$

 $\operatorname{Tr}/x + \operatorname{Neg}/x, x \rightarrow^{\rho(x)} \operatorname{Tr}/x + \operatorname{Inh}/x, x \}$

$$E_{c} := \{ Neg/x, x = ?x; Inh/x, x \oplus \tau_{e}; (Tr/x | Neg/x, x) \\ Tr/x = !x; Tr/x \oplus \tau_{d}; 0 \\ Inh/x, x = \tau_{h}; Neg/x, x \}$$

----- iteration 3 -----

$$\begin{split} \mathsf{C} &:= \{ \mathsf{Neg}/\mathsf{x}, \mathsf{x} \to^\mathsf{e} \mathsf{Tr}/\mathsf{x} + \mathsf{Neg}/\mathsf{x}, \mathsf{x} \\ & \mathsf{Tr}/\mathsf{x} \to^\mathsf{d} \mathsf{0} \\ & \mathsf{Tr}/\mathsf{x} + \mathsf{Neg}/\mathsf{x}, \mathsf{x} \to^{\mathsf{p}(\mathsf{x})} \mathsf{Tr}/\mathsf{x} + \mathsf{Inh}/\mathsf{x}, \mathsf{x} \\ & \mathsf{Inh}/\mathsf{x}, \mathsf{x} \to^\mathsf{h} \mathsf{Neg}/\mathsf{x}, \mathsf{x} \} \end{split}$$

E_c:= no change

----- termination -----

 $\begin{pmatrix} \text{Neg/x,x} \rightarrow^{\text{e}} \text{Tr/x} + \text{Neg/x,x} \\ \text{Tr/x} \rightarrow^{\text{d}} 0 \\ \text{Tr/x} + \text{Neg/x,x} \rightarrow^{\rho(x)} \text{Tr/x} + \text{Inh/x,x} \\ \text{Inh/x,x} \rightarrow^{\text{h}} \text{Neg/x,x} \\ \text{Neg/x,x} \end{pmatrix}$

From Chemistry to CGF

From Chemistry to CGF (by example)

x: $B \rightarrow^{s} A$ b: $A+B \rightarrow^{r} A+A$ c: $A+A \rightarrow^{2r} A+B$ Unique reaction names



From Chemistry to CGF (by example)

- **x:** $B \rightarrow^{s} A$ b: $A+B \rightarrow^{r} A+A$
- c: $A+A \rightarrow^{2r} A+B$



- 1: Fill the matrix by columns:
 - Degradation reaction $v_i: X \rightarrow k_i P_i$ add $\tau; P_i$ to $\langle X, v_{ii} \rangle$.

x: B →^s A b: A+B →^r A+A c: A+A →^{2r} A+B

	X _(s)	b _(r)	C _(r)
А		?;A A	
В	τ;A	!;0	

1: Fill the matrix by columns:

Degradation reaction $v_i: X \rightarrow k_i P_i$ add $\tau; P_i$ to $\langle X, v_{ii} \rangle$. Hetero reaction $v_i: X+Y \rightarrow k_i P_i$ add ?; P_i to $\langle X, v_i \rangle$ and !; 0 to $\langle Y, v_i \rangle$

- x: $B \rightarrow^{s} A$ b: $A+B \rightarrow^{r} A+A$
- c: $A+A \rightarrow^{2r} A+B$

	X _(s)	b _(r)	C _(r)
А		?;A A	?;A B !;0
В	τ;Α	!;0	

1: Fill the matrix by columns:

Degradation reaction $v_i: X \rightarrow k_i P_i$ add $\tau; P_i$ to $\langle X, v_{ij} \rangle$. Hetero reaction $v_i: X+Y \rightarrow k_i P_i$ add ?; P_i to $\langle X, v_i \rangle$ and !; 0 to $\langle Y, v_i \rangle$ Homeo reaction $v_i: X+X \rightarrow k_i P_i$ add ?; P_i and !; 0 to $\langle X, v_i \rangle$

- x: $B \rightarrow^{s} A$ b: $A+B \rightarrow^{r} A+A$
- c: $A+A \rightarrow^{2r} A+B$

1: Fill the matrix by columns:

Degradation reaction $v_i: X \rightarrow k_i P_i$ add $\tau; P_i$ to $\langle X, v_{ij} \rangle$. Hetero reaction $v_i: X+Y \rightarrow k_i P_i$ add ?; P_i to $\langle X, v_i \rangle$ and !; 0 to $\langle Y, v_i \rangle$ Homeo reaction $v_i: X+X \rightarrow k_i P_i$ add ?; P_i and !; 0 to $\langle X, v_i \rangle$

2: Read the result by rows:

$$A = ?b_{(r)}; (A | A) \oplus ?c_{(r)}; (A | B) \oplus !c_{(r)}; 0$$
$$B = \tau_{(s)}; A \oplus !b_{(r)}; 0$$

	X _(s)	b _(r)	C _(r)
А		?;A A	?;A B !;0
В	τ;Α	!;0	



- x: $B \rightarrow^{s} A$ b: $A+B \rightarrow^{r} A+A$
- c: $A+A \rightarrow^{2r} A+B$

1: Fill the matrix by columns:

Degradation reaction $v_i: X \rightarrow k_i P_i$ add $\tau; P_i$ to $\langle X, v_{ij} \rangle$. Hetero reaction $v_i: X+Y \rightarrow k_i P_i$ add ?; P_i to $\langle X, v_i \rangle$ and !; 0 to $\langle Y, v_i \rangle$ Homeo reaction $v_i: X+X \rightarrow k_i P_i$ add ?; P_i and !; 0 to $\langle X, v_i \rangle$

2: Read the result by rows:

$$A = ?b_{(r)}; A \oplus ?c_{(r)}; (A | B) \oplus !c_{(r)}; 0$$
$$B = \tau_{(s)}; A \oplus !b_{(r)}; A$$





- x: $B \rightarrow^{s} A$ b: $A+B \rightarrow^{r} A+A$
- c: $A+A \rightarrow^{2r} A+B$

1: Fill the matrix by columns:

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2: Read the result by rows:

$$A = ?b_{(r)}; A \oplus ?c_{(r)}; B \oplus !c_{(r)}; A$$
$$B = \tau_{(s)}; A \oplus !b_{(r)}; A$$





From Chemistry to Automata (by example)



Half-rate for homeo reactions



From Chemistry to CGF: Pi(C)

v: $X \rightarrow^r Y_1 + ... + Y_n + 0$ Unary Reactionv: $X_1 + X_2 \rightarrow^r Y_1 + ... + Y_n + 0$ Binary Reaction

From uniquely-labeled (v:) chemical reactions C to a CGF Pi(C):

$$\begin{array}{lll} \mathsf{Pi}(\mathsf{C}) &= & \{(\mathsf{X} = \ \oplus ((\mathsf{v}: \mathsf{X} \to^{\mathsf{k}} \mathsf{P}) \in \mathsf{C}) \ of \ (\tau_{(\mathsf{k})}; \mathsf{P}) & \oplus \\ & \oplus ((\mathsf{v}: \mathsf{X} + \mathsf{Y} \to^{\mathsf{k}} \mathsf{P}) \in \mathsf{C} \ \text{and} \ \mathsf{Y} \neq \mathsf{X}) \ of \ (?\mathsf{v}_{(\mathsf{k})}; \mathsf{P}) & \oplus \\ & \oplus ((\mathsf{v}: \mathsf{Y} + \mathsf{X} \to^{\mathsf{k}} \mathsf{P}) \in \mathsf{C} \ \text{and} \ \mathsf{Y} \neq \mathsf{X}) \ of \ (!\mathsf{v}_{(\mathsf{k})}; \mathsf{O}) & \oplus \\ & \oplus ((\mathsf{v}: \mathsf{X} + \mathsf{X} \to^{\mathsf{k}} \mathsf{P}) \in \mathsf{C}) \ of \ (?\mathsf{v}_{(\mathsf{k}/2)}; \mathsf{P} \oplus !\mathsf{v}_{(\mathsf{k}/2)}; \mathsf{O}) &) \\ & & \texttt{s.t.} \ \mathsf{X} \ \text{is a species in C} \end{array}$$



Discrete-State Semantics



Discrete Semantics of Reactions


Discrete Semantics of Reagents





CTMC

Discrete State Equivalence

- Def: 🗯 is equivalent CTMC's (isomorphic graphs with same rates).
- Thm: E 🗯 Ch(E)
- Thm: C 🗯 Pi(C)



- For each E there is an E' \approx E that is detangled (E' = Pi(Ch(E)))

Interacting Automata = Discrete Chemistry

This is enough to establish that the process algebra is really faithful to the chemistry.

But CTMC are not the "ultimate semantics" because there are still questions of when two different CTMCs are actually equivalent (e.g. "lumping").

The "ultimate semantics" of chemistry is the *Chemical Master Equation* (derivable from the Chapman-Kolmogorov equation of the CTMC).



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